

# CBE Spring 2023 Seminar Series

## Teaching Old Electrocatalysts New Tricks: Merging Concepts from Thermal Catalysis and Molecular Synthesis

**Monday,  
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10:30 AM  
EB1 1011**



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Decarbonizing the energy and chemical industries motivates the development of new catalytic technologies that use renewable energy inputs, alternative feedstocks, and spatially distributed production modalities. In this context, electrocatalysts are tasked with producing fuels, chemicals, and energy by mechanisms that fundamentally differ from those of electrolyzer and fuel cell technologies, and the thermocatalytic technologies of incumbent petrochemical processes. In this presentation I will show how concepts from thermal and molecular catalysis can stimulate new approaches for the synthesis and application of a class of heterogeneous electrocatalysts known as M-N-Cs, or metals incorporated into nitrogen-doped carbon. M-N-Cs (e.g., M = Fe, Co) catalyze electrochemical reduction of O<sub>2</sub>, such as in fuel cells, and catalyze thermochemical reduction of O<sub>2</sub> using hydroquinone (HQ) as the source of reducing equivalents. Kinetic studies reveal an unexpected mechanism for HQ-mediated O<sub>2</sub> reduction through a direct chemical pathway facilitated by a catalyst microenvironment modified by adsorbed HQ species. This alternative mechanism circumvents the rate–potential relationship observed for electrocatalytic O<sub>2</sub> reduction, opening new opportunities to design fuel cell systems that reduce O<sub>2</sub> with higher energy efficiency (i.e., lower overpotential). In a complementary effort, Fe-N-C heterogeneous catalysts were prepared to contain atomically dispersed metal active sites by adapting synthetic strategies used to metalate molecular macrocycle catalysts under solution-phase conditions and milder temperatures (150 °C) than those of conventional pyrolysis-based preparation routes (600–1100 °C). These well-defined Fe-N-C catalysts directly implicate atomically dispersed FeN<sub>x</sub> moieties as the active sites for aerobic oxidation reactions. These studies show how thermochemical and molecular concepts can be leveraged to understand and improve the structure and function of electrocatalysts that are critical for next-generation energy and chemical conversion processes.

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